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Managed grassland alters soil N dynamics and N₂O emissions in temperate steppe

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ABSTRACT

The reclamation of degraded grasslands as managed grasslands or even croplands has been 19 Q8 increasingly accelerated in recent years in China. It is well known that land use change 20 affects soil nitrogen (N) dynamics and nitrous oxide (N₂O) emissions. However, it remains 21 unclear how large-scale grassland reclamation will impact the grassland ecosystem as a 22 whole. Here, we invested the effects of the conversion from native to managed grasslands 23 on soil N dynamics and N_2O emissions by field experiments in Hulunber in northern China. 24 Soil (0–10 cm), nitrate (NO₃), ammonium (NH $^{4}_{4}$), and microbial N were measured in plots in a 25 temperate steppe (Leymus chinensis grassland) and two managed grasslands (Medicago sativa 26 and Bromus inermis grasslands) in 2011 and 2012. The results showed that conversion of 27 L. chinensis grassland to M. sativa or B. inermis grasslands decreased concentrations of NO₃-N, 28 but did not change NH₄⁺-N. Soil microbial biomass N was slightly decreased by the conversion 29 of L. chinensis grassland to M. sativa, but substantially increased by the conversion to B. inermis. 30 The conversion of L. chinensis grassland to M. sativa (i.e., a legume grass) increased N₂O 31 emissions by 26.2%, while the conversion to the B. inermis (i.e., a non-legume grass) reduced 32 N₂O emissions by 33.1%. The results also showed that conversion from native to managed 33 grasslands caused large created variations in soil NO₃-N and NH₄-N concentrations. Net N 34 mineralization rates did not change significantly in association with changes of growing 35 season and vegetation type, while variations in net nitrification rate were significantly 36 associated with changes in growing season. These results produce fresh, quantitative 37 evidence on how grassland reclamation may impact the grassland ecosystem in terms of N 38 dynamics and N₂O emissions, in persuasion of grassland protection and restoration. 39 © 2017 The Research Center for Eco-Environmental Sciences, Chinese Academy of Sciences. 40 Published by Elsevier B.V. 41

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58 Introduction

Nitrogen (N) is an important limiting nutrient for net primary
 productivity in most terrestrial ecosystems (LeBauer and

Treseder, 2008; Xu, 2003). Various diverse chemical forms of 57 soil N can be utilized by plants and microbes (Harrison et al., 58 2008; Kaštovská and Šantrůčková, 2011). For example, common 59 forms of inorganic N, such as ammonium (NH⁴₄) and nitrate 60

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(NO₃), are important N sources for plants, although organic N in 61 the form of free intact amino acids also can be taken up by 62 plants (Jones and Kielland, 2002). Soil inorganic N is mainly 09 derived from mineralization of organic matter by microbial 64 activities in natural ecosystems (Wang et al., 2005). Moreover, 65 nitrous oxide (N2O) is released during N transformation, 66 primarily via processes of denitrification and nitrification 67 (Skiba and Smith, 2000; Lan et al., 2014). N₂O is reported a **O10** 69 potent greenhouse gas with a global warming potential 298 times that of carbon dioxide (IPCC, 2013) and the largest ozone 70 depleter in the atmosphere (Reeves and Wang, 2015). Since 71 1750, the atmospheric N₂O concentration has increased by 72 about 120%, from around 270 ppb, to 324 ppb in 2011 (Wang 73 et al., 2014). 74

75The absorption and conversion of nitrogen are dependent on 76different soil nitrogen conversion processes. In nitrification, NH_4^+ is converted to nitrite (NO_2^-) and then to nitrate (NO_3^-), while 77 in denitrification, NO3 is gradually reduced to NO2 and further to 78 79 nitrogen gases (NO, N₂O, and N₂) (Wu et al., 2013). In soil, both nitrification and denitrification are performed by microorgan-80 isms (Liang et al., 2014). Land use change primarily affects plant 81 82 production, species diversity and the quality of organic or nutrient inputs into the soil due to changes in plant cover 83 84 (Raiesi and Beheshti, 2014; Zhao et al., 2007). For example, mowing of grassland was reported to increase species richness 85 86 and community stability of herbaceous plants (Collins et al., 87 1998; Yang et al., 2012), and decrease the amount of available N 88 in soil

Not only management practices, but also soil physical, 89 90 chemical and biological conditions regulate N transformation 91 (Knoepp and Swank, 2002). Soil pH, moisture content and texture are among the most important factors regulating N₂O 92 emissions (Pihlatie et al., 2004; Mørkved et al., 2007; Wang 93 94 et al., 2013). N₂O emissions increased with increasing water filled pore space (WFPS), the maximal N₂O emissions were 95 measured between 80% and 95% WFPS. (Schindlbacher et al., 96 97 2004; Qi et al., 2014).

Grasslands occupy 40% of the total land mass of China (Chen 98 et al., 2014). Natural grasslands in China are experiencing 99 degradation of different degrees (Li, 1997; Ye and Van Ranst, 100 2009). The productivity of managed grassland is several times 101 higher than that of natural grassland. As such, substantial 102 amount of degraded grasslands have been converted into 103 managed grasslands and even croplands in recent years (Liu 104et al., 2009) in order to achieve a more stable supply of higher 105106 quality forages. The area of managed grasslands expanded from 9.6 \times 10⁴ km² in 2001 to 12.8 \times 10⁴ km² in 2014, a 33% jump 107 in a period slightly longer than 10 years (National Animal 011

Husbandry, 2013). Numerous studies have been conducted to 109 investigate how land use change could affect the N cycle in 110 temperate grasslands (Pan et al., 2016). Conversion of native to 111 managed grasslands or croplands accelerates soil N minerali-112 zation (Li and Chen, 1998; Booth et al., 2005), resulting in a 8.8% 113 reduction of total soil N in the top 1 m layer of soil (Li et al., 114 2005). Whether N₂O emissions decrease nor increase after 115 conversion is a fundamental question being hotly debated 116 (Wen et al., 2005). What is clearer to date is that N losses from 117 grasslands are closely related to N availability and microbial 118 activity and thereby inherently affected by management 119 practices such as fertilization, harvesting, and tillage (Geng 120 et al., 2001; Ye et al., 2008).

In the Hulunber region, managed grasslands of legumes 122 (Medicago sativa) or non-legumes (Bromus inermis) are common 123 land use types to substitute the native temperate grasslands 124 (Leymus chinensis). Both substitute grassland types are high 125 productive of high quality forages. However, efforts to quanti- 126 tatively investigate the impacts of grassland conversion on soil 127 N dynamics and N₂O emissions are still lacking. Therefore, we 128 conducted field experiments to assess the influences of 129 grassland conversion on soil N dynamics and N₂O emissions 130 in temperate grassland ecosystems in Inner Mongolia, China. 131 We aimed to test the following hypotheses: (Attard et al., 2016) 132 land use change from native to managed grasslands promotes 133 net N mineralization; (Bernard et al., 2007) conversion to the 134 legume M. sativa increases N₂O emissions because biological N 135 fixation provides more NH₄⁺; and (Bedard-Haughn et al., 2006) 136 soil moisture has stronger impacts on N2O emissions than 137 temperature in this semiarid region. 138

1. Materials and methods

1.1. Research site

This study was conducted at the Hulunber Grassland Ecosystem 142 Observation and Research Station (49°19'35″N, 119°56'52″) in 143 north-eastern Inner Mongolia in China. The frost-free period is 144 110 days, and the average annual precipitation was totaled at 145 320 mm during the year of 2000–2010, 80% of which fell between 146 July and September. Monthly mean temperature varied between 147 the maximum of 36.2°C in July and the minimum of –48.5°C in 148 January. Chestnut soil is the main soil type (Table 1). 149

Experiments were conducted on two managed grasslands 150 (M. sativa and B. inermis) and a native grassland (L. chinensis). 151 The plots were arranged in a split-plot design with land use as 152 the main treatment. Each 10×10 m plot was consisted of 153

t' Q1	Table 1 – Site characteristics and soil properties for the different grassland sites evaluated during the growing season.						
t1.3 t1.4	Grassland type	M. sativa	B. inermis	L. chinensis			
t1.5	pH value	6.96 ± 0.68a	6.88 ± 1.05a	6.72 ± 0.98a			
t1.6	Alkali hydrolyzable nitrogen (mg/kg)	135.57 ± 5.79a	124.37 ± 8.36b	125.42 ± 5.55b			
t1.7	Available potassium (mg/kg)	307.88 ± 21.87a	205.81 ± 67.97c	235.49 ± 34.45b			
t1.8	Total nitrogen (g/kg)	2.40 ± 0.67a	2.29 ± 0.24a	2.31 ± 0.15a			
t1.9	Soil organic carbon (g/kg)	16.17 ± 3.01a	15.83 ± 2.65b	16.02 ± 2.76b			
t1.10	Texture	Sandy soil	Sandy soil	Sandy soil			
t1.11	Soil depth (cm)	10	10	10			
t1.12	Average aboveground biomass (kg/ha)	5843.36 ± 250.17a	$6044.91 \pm 309.05a$	3044.85 ± 167.16b			

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three subplots of 15 m². The two managed grasslands were 154established by converting the original temperate steppe 155(L. chinensis) to M. sativa or B. inermis through re-seeding in 1562009. Samples were collected from May to September on the 157third and fourth year after planting. The seeding amounts of the 158M. sativa and B. inermis grasslands were 15.0 and 22.5 kg/ha, 159respectively. The plowing depth was 2 cm in both grasslands. 160 161 Before sowing, 2,4-D butyl ester was applied as subsurface

162 treatment to clear weeds. No fertilizer was applied.

163 1.2. Meteorological data

164 Meteorological data was collected by an automatic meteo-165 station, located 100 m from the experimental plots.

166 1.3. Net N mineralization

Many researches revealed that for most of the arid and semiarid 167grasslands, N₂O was predominantly produced by microbial 168nitrification (Cookson et al., 2006; Xu et al., 2008; Zhong et al., 1692014). Heterotrophic nitrification reported to be the dominant 170process of N₂O production in Inner Mongolia steppe, while the 171 contribution of denitrification seemed not significant (Du et al., 172173 2001; Du, 2006). Soil N mineralization and nitrification rates 174 were measured using in situ incubations in May 2011. At the 175start of each incubation period, soil cores were taken in pairs 012 using PVC tubes of 10 cm in height and 5 cm in diameter from three randomly chosen positions in each plot. The litter layer 177 was removed before sample was taken. One of each pair of cores 178(initial sample) was removed and sent to the laboratory in an 179icebox to determine the initial soil ammonium (NH₄⁺-N) and 180 nitrate (NO₃-N) concentrations. The second core (incubation 181 sample) was wrapped with low-density polyethene on the top 182 and with gauze under the bottom allowing gas movement but 183 preventing leaching. The core was then returned to its original 184 position. During the incubation period, the samples were 185collected every 30 days from June to September in 2011 and 186 2012. The amount of mineral N accumulated in the sample was 187 determined. Within each core, changes in inorganic N content 188 during the incubation period represent the net N mineralized 189from the organic sources. 190

191 All collected soil cores were kept cool before reaching the 192laboratory and stored in a refrigerator at 4°C in laboratory until analyzed, usually within 36 hr. In the laboratory, the PVC 193tubes were removed and each soil core was well mixed by 194hand, excluding roots and stones. A 5 g sub-sample was 195extracted in 20 mL of 2 mol/L KCl for 1 hr, filtered through 196Whatman 42 filter paper and centrifugal oscillated for 30 min. 197The NH₄⁺–N and NO₃⁻–N contents were then analyzed with an 198 ultraviolet spectrophotometer (752PC, Shanghai, China). 199

The inorganic N concentrations were expressed on a dry weight basis. To obtain oven-dry weight, a 20 g sub-sample was placed in a 105°C oven for >12 hr.

203 Rates of net N mineralization and nitrification were 204 calculated using the following equations:

$$N_{min} = ([NH_4^+] + [NO_3^-])_j - ([NH_4^+] + [NO_3^-])_i / t_{(j-i)}$$

where N_{min} is the net N mineralization rate(μ g N dw (dry weight)/(g·day)), [NH₄⁺] is the NH₄⁺ concentration (μ g N dw/g),

 $[NO_3^-]$ is the NO_3^- concentration (µg N dw/g), *j* is the soil 208 incubation *j* days, *i* is the soil incubation *j* days, and $t_{(j-i)}$ is the 209 soil incubation from ith to *j*th day. The net N nitrification rate 210 (N_{nit}, µg N dw/(g·day)) is given by: 211

$$\mathbf{N}_{\text{nit}} = \left([\mathbf{NO}_3^-]_j - [\mathbf{NO}_3^-]_i \right) / t_{(j-i)}.$$

1.4. Microbial biomass N (MBN) Q13

Fresh soil samples were collected every month during growing 215 season from June to September in 2011 and 2012 using an 216 earth-boring auger. Three points were randomly selected in 217 each plot, and three soil samples were taken at 0–10 cm depth. 218 Soil samples were put into plastic bags, and immediately 219 brought to the laboratory for measurements of MBN and soil 220 moisture content. 221

The MBN was determined by the chloroform fumigation 222 direct extraction method. Paired samples of moist soil of 10 g 223 in weight were placed in 120 mL glass cups. One sample of the 224 pair was extracted with 50 mL of 0.5 mol/L K₂SO₄ by shaking 225 for 1 hr, and the resulting suspension was filtered. The second 226 sample was fumigated with ethanol-free chloroform for 24 hr 227 in the dark at 22°C. The chloroform was then removed and the 228 sample was extracted with 0.5 mol/L K₂SO₄ as described 229 above. The MBN was calculated as the difference in dissolved 230 organic N between the fumigated and non-fumigated extracts 231 using the following equation: 232

 $MBN = (N_{fumigated} - N_{non-fumigated})/0.45$

where $N_{fumigated}$ is the total dissolved organic N extracted 233 after fumigated with chloroform for 24 hr, and $N_{non-fumigated}$ 235 is the total dissolved organic N directly extracted without 236 fumigation. 237

1.5. Soil properties 238

Bulk density was calculated as the quotient between the dry 239 weight and the core volume. Soil pH was measured using a pH 240 probe (SG2, Beijing, China) in 1:2 soil water solution of 10 g dry 241 soil in 20 mL water. Total N content was measured by dry 242 combustion on a LECO CNS-1000 elemental analyzer (LECO 243 Corporation, St. Joseph, MI, USA) (Du and Gao, 2006). 244

1.6. Measurement of N₂O emissions

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Nitrous oxide emissions were measured using the dark 246 static closed chamber method from June to September in 247 2011 and 2012, respectively. The chamber volume was 0.2 m^3 248 (0.5 m × 0.5 m × 0.8 m). The chamber collars were installed 249 into the soil in June 2011 to a soil depth of 10 cm. The N₂O 250 emissions were typically measured at local time from 9:00 to 251 10:00 (demonstrated greenhouse gas fluxes measured during 252 this time period are representative of the daily mean flux) on 253 each sampling day and 1–2 times per week during the growing 254 season (i.e., June through September) (Xu et al., 2015). Gas **Q14** samples were taken at 0, 10, 20, 30 and 40 min after chamber 256 installation using 50 mL polypropylene syringes equipped with 257 three-way stopcocks. After sampling, N₂O flux was analyzed 258

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using a gas chromatograph (Chemical Engineering Department 259Dalian bright chemical industry research Institute, Dalian, 260China) equipped with a ${}^{63}N_i$ electron capture detector (ECD) 261 operated at 300°C. The fluxes were calculated based on the 262assumption of linear concentration increase inside of a 263chamber to minimize the negative effect of chamber closure 264on N₂O production (Wang et al., 2014). Prior report showed that 265the increase in N₂O concentrations remained linear for up to 2662672 hr following chamber closure, the coefficients of determination (R^2) of the linear regression were greater than 0.98 268(p < 0.001) (Zhang and Wang, 2008). Q15

The N₂O emissions were calculated as the slope of the linear regression of the N₂O concentration change over time.

The N₂O emissions inside the chambers were estimated using the following equation:

$$\begin{split} F &= [M \times V \times (\text{C2} \times 273 \times \text{P2})/(\text{P0} \times \text{T2}) - \text{C1} \times 273 \times \text{P1}/(\text{P0} \times \text{T1})]/\text{A} \\ &\times (\text{T2-T1}) \times 22.4 \end{split}$$

where, F (μ g N₂O–N/(m²·day)) is the N₂O flux; M (g/mol) is the 275 molecular weight of N₂O–N; V (0.2 m^3) is the volume of the 276 chamber; A (0.25 m^2) is the area from which N₂O was emitted 277 into the chamber; T1 and T2 (day) are the sampling and end 278 times, respectively; C1 and C2 are the concentrations of N₂O 279 (cm³/m³) at the beginning and end of gas collection, respec-280 tively; T1 and T2 are the soil absolute temperatures (K) at the 281 282 beginning and end of gas collection, respectively; P1 and P2 283 are the atmospheric pressures inside the chamber (Pa) at the 284 beginning and end of gas collection, respectively; and PO is the 285 standard atmospheric pressure at 273 K (Pa). We assumed 286 that P2 = P1 = P0.

The cumulative N_2O fluxes were calculated by interpolating 287 the N_2O fluxes that were measured during the sampling periods 288 (Dong et al., 2000; Peng et al., 2011). Q16Q17

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1.7. Statistical analyses

Correlation analysis was used to evaluate the relationships 291 between N₂O emissions and temperature, soil moisture content, 292 NO₃–N, NH₄⁺–N, MBN, N mineralization rate and nitrification rate. 293 Models were fit after data were tested for normality. Analyses 294 were conducted using SAS V9.2 (SAS Institute Inc., Cary, NC, USA). 295 For multiple linear regression analysis, stepwise variable selec- 296 tion was used in most cases. Principal component analysis (PCA) 297 was used to determine the factors influencing N₂O emissions, 298 and analysis of variance was used to analyze the differences in 299 N₂O emissions between species and months of the year. 300

2. Results

2.1. Meteorological conditions

The average temperature of experimental site of 2011 and 2012 304 was -1.60 and -1.86°C, respectively, lower than the average 305 annual temperature of 0.25°C between 2000 and 2010. Precipi-306 tation in 2011 was 317.5 mm and lower than the average 307 precipitation of 323 mm during, while precipitation in 2012 308 (i.e., 320 mm) was similar to the average level. Fig. 1 shows the 309 average daily temperature and precipitation in 2011 and 2012. 310 During the field observation period from June to September in 311 each year, the daily average temperatures were between 20 and 312



Fig. 1 - Seasonal changes of air temperature and precipitation in the three different types of grassland.

313 30°C and the total precipitation accounted for about 70% of the314 total annual precipitation.

315 2.2. Dynamics of NH⁺₄–N, NO⁻₃–N and MBN

The NO₃-N concentrations were higher and more varied 316 between the three grassland types in 2011 than in 2012. In 317 2011, the NO₃-N concentrations in the managed grasslands 318319 were 19.0%-21.4% lower than those in the native grassland $(31.04 \pm 6.74 \mu g N dw/g)$ (Fig. 2a). The NH₄⁺–N concentrations 320 in the native grassland (22.63 \pm 4.38 μ g N dw/g) were 1.07 and 321 1.10 times higher than those in the M. sativa grassland and 322 B. inermis grassland, respectively. In 2011, the NO₃-N concen-323 trations in the native grassland (13.91 \pm 4.8 μ g N dw/g) were 324 0.90 and 1.76 times higher than those in the M. sativa grassland 325 and B. inermis grassland, respectively during the growth season. 326 In 2012, the NH₄⁺-N concentrations in the native grassland 327 $(12.31 \pm 2.86 \mu g N dw/g)$ were slightly lower than those in the 328 managed grassland (Fig. 2b). 329

The ratios of NH₄⁺–N to NO₃⁻–N in the three grassland types 330 were only higher than 1 in August and October, 2011. The NH₄⁺– 331 N/NO₃⁻–N ratios ranged between 0.26 and 2.44 for the native 332 *L. chinensis* grassland, between 0.12 and 3.24 for *M. sativa*, and 333 between 0.34 and 16.61 for *B. inermis*. In June, July and 334 September, NO₃⁻ dominated the inorganic N and accounted for 335 64.4%–82.3% of total inorganic N. In August and October, NH₄⁺–N 336 dominated the inorganic N and accounted for 56.4%–79.2% of 337 total inorganic N in 2011. From July to September in 2012, NH₄⁺–N 338 accounted for 56.1%–94.3% of the total inorganic N. There was a 339 significant correlation between the NO₃⁻–N concentrations and 340 the vegetation type or the growing season (p < 0.05), but as for 341 NH₄⁺–N, the only significant correlation was found with the 342 growing season (p < 0.05) (Table 2).

The MBN values greatly varied across the three grassland 344 types during the whole growing season (p < 0.05) (Fig. 2c, 345 Table 2). The MBN values were much higher in 2012 than 2011. 346 The MBN values in the M. sativa grassland were 17.01%–41.57% 347 higher than those in the native grassland (66.38 ± 1.01 and 348



Fig. 2 – Seasonal dynamics of (a) soil nitrate (NO_3^--N), (b) soil ammonium (NH_4^+-N), and (c) soil microbial biomass nitrogen (MBN). The values indicate average ± standard error of three replicates. Different lowercase letters indicate significant differences between grassland types for each month (p < 0.05).

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Index	Source	DF	Mean square	F value	р
NO ₃	Growing season	4	394.84	9.80	<0.0
	Туре	2	238.99	5.93	0.00
	Growing season × Type	8	28.94	0.72	0.67
NH_4^+	Growing season	4	287.54	38.67	<0.0
	Туре	2	4.67	0.63	0.54
	Growing season × Type	8	14.76	1.98	0.08
MBN	Growing season	4	32,897.91	2982.75	<0.0
	Туре	2	23,702.63	2149.05	<0.0
	Growing season × Type	8	4571.87	414.52	<0.0
N ₂ O	Growing season	3	0.00029	0.83	0.49
	Туре	2	0.00075	2.14	0.14
	Growing season × Type	6	0.0007	1.99	0.11
N _{min}	Growing season	4	0.05	0.80	0.53
	Туре	2	0.09	1.38	0.27
	Growing season × Type	8	0.12	1.82	0.11
N _{nit}	Growing season	4	1.79	7.24	0.09
	Туре	2	0.95	3.85	0.14
	Growing season × Type	8	0.56	2.28	0.00

t2.23 DF means degrees of freedom, microbial biomass nitrogen, and soil moisture.

t2.23 Numbers in bold indicate significant difference at p < 0.05.

t2.28 MBN: microbial biomass nitrogen.

109.15 ± 9.59 mg/g in 2011 and 2012, respectively). In contrast, 349 the MBN values in the B. inermis grassland were 11.15%-30.09% 350

lower than those in the native grassland (Fig. 2c). 351

2.3. Net mineralization and nitrification 352

353 Net mineralization rates varied over the growing season from June to October in each year, and the overall patterns of change 354 were similar in both years (Fig. 3). During the period from June to 355 August, net mineralization rates increased from -0.93 ± 0.04 356 357 to $0.99 \pm 0.15 \,\mu g \,\text{N} \, \text{dw}/(g \cdot \text{day})$ in L. chinensis grassland, from -0.51 ± 0.07 to 0.74 ± 0.11 µg N dw/(g day) in M. sativa grass-358 land, and from -0.84 ± 0.1 to $0.91 \pm 0.23 \mu g N dw/(g \cdot day)$ in 359 B. inermis grassland in 2011 (Fig. 3a). The majority of the positive 360 and negative peaks occurred in August and September in 2011. 361 There was no significant correlation between net mineraliza-362 tion rate, vegetation type and plant growing season (p > 0.05). 363 The net mineralization rates did not change significantly with 364 the growing season (p > 0.05) (Table 2). 365

Compared with the net mineralization rates, the soil net 366 nitrification rates were much lower in magnitude and smoother 367 368 in variation over time. The net nitrification rate in L. chinensis grassland was only positive in July and September, 2011, and in 369 June and October, 2012. The maximum net nitrification rate was 370 $0.72 \pm 0.60 \mu g N dw/(g day)$ in October 2011. The net nitrification 371 372 rates in M. sativa grassland were positive except in August and October, 2011 and in July and August, 2012. The maximum 373 net nitrification rate in M. sativa grassland was 0.53 ± 3740.48 µg N dw/(g·day) in October 2012. The maximum net nitri-375 fication rates in B. inermis grassland were positive only in July 376 and September, 2011 and June and October, 2012, with negative 377 values for all other periods. The maximum net nitrification rate 378 in B. inermis grassland was 0.43 \pm 0.36 μg N dw/(g·day) in June 379 380 2011. The peak net nitrification rates in L. chinensis grassland 381 occurred in July 2011 and October, 2012, and the lowest rates were measured in October 2011 and July, 2012. The highest net 382

nitrification rate in M. sativa grassland occurred in September 383 2011 and October 2012, the lowest rates occurred in October 2011 384 and July 2012. The highest net nitrification rate in B. inermis 385 grassland was measured in July 2011 and October 2012, and the 386 lowest rates occurred in October 2011 and July 2012 (Fig. 3b). The 387 positive values of net nitrification rate indicated that NO₃-N 388 produced by soil nitrification remained as residual NO3 in 389 soil after immobilization by microbes. There was a significant 390 correlation (p < 0.05) between net nitrification rate, vegetation 391 type, and growing season (Table 2). 392

2.4. Nitrous oxide emissions

N₂O emissions varied greatly across treatments. The average 394 N₂O emissions showed similar patterns across the three types 395 of grasslands with the highest emissions observed during late 396 June and July and lower values thereafter (Fig. 4). The N₂O 397 internal fluctuation was clear in the L. chinensis grassland. The 398 seasonal pattern of N2O emissions in the M. sativa grassland was 399 similar to that in the L. chinensis grassland, but with significantly 400 lower (p < 0.05) emissions rates in 2011 (85.4 \pm 13.12 to 1746.3 \pm 401 35.47 μ g N₂O–N/(m²·day)). The N₂O emissions showed a net 402 positive flux throughout the growing season (272.6 \pm 39.46–403 3931.8 \pm 101.37 µg N₂O–N/(m²·day)), exhibiting relatively large 404 fluctuations in the early stage of the wet summer and autumn 405 in 2012. Emissions from B. inermis grassland were relatively low 406 in 2011, but fluctuated greatly in early growing season in 2012 407 (Fig. 4). The flux rates ranged between 129.7 ± 30.06 and 2847.4 ± 408 291.08 μ g N₂O–N/(m²·day) during the whole season. 409

Overall, the average rates of N₂O emissions were different 410 between grassland types from 2011 to 2012, decreasing in 411 the order M. sativa $(1038.7 \pm 89.19 \,\mu g \, N_2 O - N/(m^2 \cdot day)) > L. 412$ chinensis (823.3 ± 46.35 μ g N₂O–N/(m²·day)) > B. inermis (695.0 ± 413 78.29 μ g N₂O–N/(m²·day)). Compared with the L. chinensis grass- 414 land, the average N₂O emissions of the M. sativa grassland were 415 26.2% higher, while emissions of the B. inermis grassland were 416

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Fig. 3 – Seasonal dynamics of net mineralization rate and nitrification rate in different grassland types. The values are average ± standard error of three replicates.

33.1% lower. However, differences in N₂O emissions between
 growing seasons and vegetation types were found insignificant

419 (p > 0.05; Table 2).

420 **2.5.** Correlation analysis between N₂O emissions and 421 environmental factors

The influence of environmental factors on N_2O emissions was tested using correlation and PCA analyses. Correlations between N_2O and NO_3 -N or nitrification rate (N_{nin}), temperature were found positively significant (p < 0.01) (Table 3, Fig. 5). 425 Positive correlations were also found between N₂O and MBN, 426 net mineralization rate or net nitrification rate (p < 0.05). There 427 was a highly significant negative correlation between N₂O and 428 NH₄⁺–N (p < 0.01). However, no correlation was found between 429 N₂O and precipitation or soil moisture (p > 0.05). The PCA 430 results (Fig. 5) indicated that NO₃⁻–N, MBN, and N_{nit} were the 431 main factors influencing N₂O emissions. Among them, NO₃⁻N, Q18 N_{nin} or MBN was the most significant, followed by NH₄⁺–N, N_{min} 433 and temperature. 434



Fig. 4 – Seasonal dynamics of N_2O emissions in different grassland types. The values are average \pm standard error of three replicates.

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t3.1 t3.2	Table 3 – Mu nitrogen, an	ultiple linear regression analysis between N2O emission and NH4, NO3, N _{nin} , air temperature, microbial biomass nd soil moisture.					
$t_{3.3}^{43.3}$	Species	Multiple linear regression analysis	R ²	p value			
t3.5	L. chinensis	N_2O emission = (6.23 ± 1.05) + (20.5 ± 2.45) × $[NO_3^-]$ + (4.67 ± 0.35) × N_{nin} – (9.68 ± 1.25) × Moisture	0.47	0.02			
t3.6	M. sativa	N_2O emission = (-22.47 ± 9.8) + (3.1 ± 0.6) × $[NO_3^-]$ + (5.78 ± 0.22) × N_{nin} + (38.7 ± 13.6) × N_{nit}	0.40	0.03			
t3.7	B. inermis	N_2O emission = (5.65 ± 3.6) + (6.3 ± 1.4) × $[NO_3]$ + +(6.12 ± 0.27) × N_{nin} + (19.6 ± 7.9) × N_{min}	0.55	0.04			

436 3. Discussion

437 3.1. Effects of land use change on soil N dynamics

Anthropogenic factors play a major role in atmospheric N 438 deposition into natural soil systems, while human activities 439 exert both intentional and unintentional alterations (Houlton 440 et al., 2013). From a global budget perspective, agricultural 441 emissions of NOx or NH3 comprise a large-scale recycling, 442 representing a new N input to downwind ecosystems 443 (Houlton and Morford, 2015; Zhou et al., 2016). Land use 444 change often alters soil physical properties such as bulk 445 density and porosity, which affect soil N mineralization via 446 447 effects on microbial activities (Gardner and Drinkwater, 2009; 448 Meagan and Laurie, 2012). In the current study, mineralization 449 was increased by land use change (Fig. 2), which causes 450 significant changes in soil NO₃-N and NH₄-N concentrations. 451 This may be due to increased aeration and porosity in managed soils following plowing (Tibisay et al., 2007). The 452conversion from native grasslands to managed grasslands 453may also increase the aboveground biomass (Table 2). 454 Different ecosystems or species show differential preferences 455 for nitrogen (Oliveira et al., 2016; Gerschlauer et al., 2016; 456Macduff et al., 1997; Attard et al., 2016). Legumes are 457 considered a high quality forage due to their high protein 458content (Mortenson et al., 2004). More importantly, it is well 459known that legumes can increase N availability in soils by 460

fixing atmospheric N in an available form (Spehn et al., 2002). 461 Increases in soil total N and available N concentrations were 462 observed when cropping systems involved more legume crops 463 (Li et al., 2015). Changes in soil NO_3^-N and NH_4^+-N were also 464 reflected in plant N accumulation, suggesting responses to 465 enhanced soil N mineralization (Bernard et al., 2007; Davies 466 et al., 2001; Nevens and Reheul, 2002). In this paper, the higher 467 NO_3^-N concentrations observed in 2011 than in 2012 may be 468 related to rainfall patterns; the earlier rainfall supplied suffi-469 cient water for soil in 2011, which promoted N transformations 470 (Fig. 1). Booth et al. (2005) suggested that the temperature– 471nitrification relationship depends in part on the availability of 472 NH_4^+ .

Land use change influences nitrification and consequently 474 NO_3^--N production in grasslands (Fig. 2). Moreover, soil NO_3^--N 475 concentrations in native grasslands are higher than in managed 476 grasslands, indicating that native grasslands have higher 477 nitrification rates and release more NO_3^--N than managed 478 grasslands. Burton et al. (2007) reported that it is possible that 479 the conversion from a mixed species forest to a single species 480 forest has changed the quality of organic matter input and 481 subsequently microbial population and diversity, which has 482 ultimately resulted in higher nitrification rates in the native 483 forest soils compared to the plantation soils, the truth could 484 analogized to grassland appropriately. Moreover, the high 485 microbial N demand in native grassland and consequently low 486 substrate (NH_4^+) availability for nitrification and most of the 487 NO_3^--N produced was probably taken up by cultivated forages or 488



Fig. 5 - Principal component analysis between N₂O emissions and their related impact factors.

lost by high NO₃-N leaching may also contribute to low NO₃-N 489 concentration in the managed soils (Xu and Xu, 2015). Mean-490 while, we only collected 2 year data, we will continue to do the 491 experiment to reveal the relative regular in further in the 492following days. Given the absence of plant uptake and leaching 493 in the incubation cores, NO3 immobilization and reduction were 494 likely the dominant NO3 consumption pathways. Nitrogen 495 producing and consuming processes may be simultaneously 496497 activated by enzymes when the intracellular N content of associated microorganisms reaches critically low levels 498 (Bedard-Haughn et al., 2006; Bengtson et al., 2005). At the early 499stages of litter decomposition, microbes immobilize N from the 500soil to facilitate their own growth, whereas at later stages of 501decomposition when the C:N ratio of the litter materials has 502substantially decreased, N is released to the soil (Cookson et al., 5032007). Warming can stimulate net N mineralization under 504optimal moisture conditions and thereby leads to microbial 505growth and increased secretion of NH₄⁺-N. We found that MBN 506 content reached peak levels in August, perhaps as a result of 507this process (Fig. 2c). Because the soil columns were incubated 508under aerobic conditions, the relatively higher pH in native 509grassland soil than in managed grassland soil (Table 1) may 510have promoted nitrification in the native grassland. Such 511 512differences in nitrification activities between different land use types were also observed elsewhere, (e.g., Sun et al., 2013). 019

514 3.2. Effects of land use change on N₂O emissions

Previous research suggested that land use change may cause 515N₂O emissions to increase (MacDonald et al., 2011; Velthof 516et al., 2010). In this paper, we found that the conversion to 517newly-planted M. sativa resulted in higher emissions, but the 518conversion to B. inermis resulted in lower N2O emissions than 519the native grasslands (Fig. 3). This confirms our second 520hypothesis that M. sativa grassland increases N₂O emissions 521because biological N fixation promotes N inputs into the soil. 522Although increased N₂O emissions are often attributed to soil 523disturbances and substrate incorporation by tillage, the differ-524ence between results obtained from M. sativa and B. inermis plots 525suggests that the effects of land use change on N₂O emissions 526are more complex than previously stated. 527

528When mineralization is increased by soil temperature, nitrification will also be increased, as long as no other NH₄-529assimilating or consuming processes dominate (Bedard-530Haughn et al., 2006). Chu and Grogan (2010) indicated that **O20** nitrification was not limited by NH₄⁺ availability alone; 532however, we found no significant correlation between N₂O 533and NH⁺₄. The origin of N₂O emissions from nitrification is still 534on debate. Ritchie and Nicholas (1972) suggested that NH₄⁺ 535oxidizers reduced NO2 to N2O to minimize intracellular 536537accumulation of NO₂ which is toxic. Remde and Conrad (1990) showed that N₂O could originate from nitrite produced 538inside the cells. Khalila et al. (2004) showed that N₂O 021 production during nitrification was usually very low. 540

Many studies have suggested that temperature and soil moisture played an important role in N₂O emissions (Liu et al., 2015; Lu et al., 2015; Wang et al., 2016). Our results were largely in line with this finding. No significant linear correlation was found between N₂O emissions and soil moisture (p > 0.05), whereas a significantly positive correlation was found between N_2O emissions and temperature. This indicates that soil 547 moisture conditions are not a limiting factor on N_2O emissions 548 during the growing season in temperate grassland in Hulunber, 549 showing conflicts with our third hypothesis. 550

Zheng et al. (2002) and Houlton and Morford (2015) suggested 551 that nitrification dominated N2O emissions when air tempera- 552 ture was between 15 and 35°C. However, when temperature was 553 lower than 5°C or higher than 40°C, nitrification may be 554 inhibited. During the entire growth season in Hulunber, the 555 atmospheric temperature was between 15 and 25°C, with an 556 average temperature of 18.3°C. This indicates that nitrification 557 could be responsible for N_2O emissions at our research site. A 558 significantly positive correlation between N2O emissions and 559 soil NO₃-N ($\mathbb{R}^2 = 0.56$, p < 0.05) suggests that nitrification is a 560 dominant process here, otherwise the higher NO₃ concentra- 561 tions as substrate for denitrification will lead to higher denitri- 562 fication. However, Jiao et al. (2015) found that soil temperature 563 had no effect on N₂O emissions ($\mathbb{R}^2 = 0.003$, p > 0.05). Further- 564 more, Du et al. (2000) and Xu (2003) noted that N₂O emissions in 022 Inner Mongolia grasslands were not sensitive to temperature 566 changes because soil moisture may counterbalance the effect of 567 soil temperature on N₂O emissions. It is clear that further 568 research is needed on this important subject matter of N2O 569 emissions by involving, among other factors, soil temperature 570 and soil moisture. 571

4. Conclusions

Conversion of L. chinensis grasslands to M. sativa or B. inermis 574 grasslands lowered the concentration of soil NO₃-N, but caused 575 little change in NH4-N. MBN decreased with conversion of 576 L. chinensis grasslands to M. sativa grasslands, but to the contrary, 577 conversion of L. chinensis grassland to B. inermis grasslands 578 strongly raised MBN. Conversion of native grasslands to 579 managed grasslands influenced variations in the NO3-N and 580 NH⁺-N concentrations. Net mineralization rates did not change 581 in growing seasons or vegetation type. However, the growing 582 season had a significant impact on net nitrification rate, and 583 thus elevated N₂O emissions in M. sativa grasslands. In contrast, 584 conversion to B. inermis grasslands lowered N2O emissions. The 585 effects of land use change on N2O emissions depend on the type 586 of managed grasslands. Conversion from native grasslands to 587 legume pastures caused 26.2% increase in N2O emissions, while 588 conversion to the non-legume grasslands reduced N₂O emis- 589 sions by 33.1%. These results provide new insights into the 590 mechanisms regulating the impact of grassland land use change 591 on nitrogen conversion, and potentially facilitate the protection 592 and restoration of the grassland ecosystems. 593

Uncited reference

Wakelin et al., 2009

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